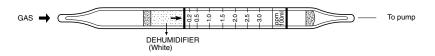
HYDROGEN SULPHIDE



1. PERFORMANCE

4) Shelf life : 2 years 5) Operating temperature : $0 \sim 40 \,^{\circ}\text{C}$

6) Reading : Direct reading from the scale calibrated by 1 pump stroke

7) Colour change : Pale yellow→Pink

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 5% RSD-high: 5%

3. CHEMICAL REACTION

By reacting with Mercury chloride (II), Hydrogen chloride is produced and PH indicator is discoloured. H2S + HgCI2→HgS + HCI

4. CALIBRATION OF THE TUBE

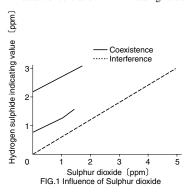
PERMEATION TUBE METHOD

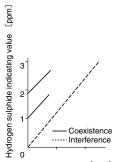
5. INTERFERENCE AND CROSS SENSITIVITY

Substance	Interference	Coexistence
Substance	IIIIGHEIGHG	COEXISIENCE
Sulphur dioxide	Whole reagent is changed to Pale red, but Purplish red stain is indicated H ₂ S	
FIG.1	concentration.	
Hydrogen selenide	Similar stain is produced.	Higher readings are given.
Arsine	"	"
Mercaptans FIG.2	"	"
Phosphine	"	"
Hydrogen cyanide	Whole reagent is changed to Red.	"

(NOTE)

In case of 1/2 pump strokes, following formula is available for the actual concentration. Actual concentration $= 2 \times \text{Reading value}$





Ethyl mercaptan (ppm) FIG.2 Influence of Ethyl mercaptan